REFERENCES

- Nagarathnam, A. K. and Rajamani, T. S., 1963, Studies on fruit setting in chillies. *Madras agric. J.*, 50: 138-139.
- Nowak, T. J., 1980, Effect of foliar fertilisation of Capsicum annuum L. in hydroponics culture on fruit yield and its capsaicin content. Acta Agrobotonica 33 (1): 59-79 (Hort. Abst, 51: 5508)
- Patil, P. K. and Ballal, A. L., 1980, Effect of seed treatment and foliar application of various plant growth regulators on flower

- drop and yield of chilli var. NP 46-A. J. Maharastra Agri. Univ., 5(3): 195-197.
- Patil, U. B., Sangale, P. B. and Desai, P. B., 1985, Chemical regulation of yield and composition of chilli (Capsicum annuum L.)-Curr. Res Reporter, 1: 39-43.
- Rajendraprasad, A. and Subramaniam, K., 1978, Response of chillies to soil and foliar application of nitrogen phosphorous and potassium, Andhra Agric. J. 25: 95-101.

Karnataka J. Agric. Sci., 2 (4): (332-334): 1989

Comparing Methods of CEC Estimation for Vertisols

In vertisols, which have higher clay content, estimation of CEC by the method suggested by Bower et al. (1952) may not give correct values; because of dispersion colloidal material in organic solvents. Therefore, a CEC estimation method suggested by Gupta et al. (1985), which eliminates the use of organic solvents for removing excess salts was compared with the commonly followed method of Bower

et al. (1952) for CEC estimation of somevertisols of Karnataka.

In the method of Gupta et al. (1985), soils were sodium saturated with 0.1 N Na OAC + 0.4 N NaCl solution, adjusted to pH 10.0 and then Na was displaced by 1 N Mg (NO₈)₂ solution adjusted to pH 8 6. The leachates were analysed for Na and Cl. Cation exchange capacity was calculated using the following equation.

Whereas, in the method of Bower et al. (1952), soils were sodium staturated with 1 N NaOAc solution adjusted to pH 8.2. The soils were washed three times with alcohol to remove the excess salts. Then adsorbed Na was extracted in 1 N NH, OAC solution.

Exchangeable Na and K of these soils and chloride in the leachate were determined by the methods described by Richard (1968).

The data on cation exchange capacity (CEC), exchangeable sodium percentage

(ESP), exchangeable potash percentage (EPP), time required for CEC estimation and cost of chemicals used for CEC estimation, by two methods compared, are given in Table 1.

The CEC values obtained by Bower et al. (1952) method were higher by 19-48 per cent over those by Gupta et al. (1985) method. The high CEC values obtained by Bower et al. (1952) method are apparently due to the use of alcohol for washing excess salts. Higher Na values noticed, which in turn gave higher CEC values may be due to extraction of retained soluble sodium salts if washing was incomplete (Gupta et al., 1985). The ESP and EPP values calculated with CEC values by Bower et al. (1952) method were lower by 16-32 per cent compared to those calcu-

lated with CEC values by Gupta et al. (1985) method. This was due to the higher CEC values obtained in the former method. This means Bower et al. (1952) method under estimates the ESP and EPP values, which is not desirable.

The time required for CEC estimation of six soil samples by Bower et al. (1952) method was more by about 75 minutes compared to that by Gupta et al. (1985) method. The cost of chemicals required for CEC estimation of 10 soil samples by Bower et al. (1952) method was about six times of that for CEC estimation by Gupta et al. (1985) method (Table 1).

Looking to the advantages in time factor and cost of chemicals involved and

Table 1. CEC, ESP, EPP values, time required and cost of chemicals as affected by methods of CEC estimation in some vertisols

Soil series	CEC by Bower et al. (1952) method			CEC by Gupta et al. (1985) method		
	CEC (m.e./100 g)	ESP %	EPP %	CEC (m.e./100 g)	ESP %	EPP %
Hanchinal	81.53	1.34	1.37	57.07	1.91	1.96
Kiresur	81.52	1.94	1.41	55.20	2.86	2.08
Mangoli	76.08	18.0	1.30	63.91	0.97	1.55
Durdundi	64.13	0.69	1.85	49.67	0.89	2.38
Mean	75.82	1.20	1.48	56.46	1.66	1.99
Time required for CE	EC	-			· · ·	
samples (minutes)	375		•	300		
Cost of chemicals use for CEC estimation of						
10 soil samples (Rupees) 63.91			10.49			

the fact that Gupta et al. (1985) method eliminates the use of alcohol and thus the sources that produce errors in the method

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of Bower et al. (1952), the adoption of former method seems to be appropriate for CEC estimation of vertisols.

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REFERENCES

Bower, C. A., Reitemeier, R. F. and Fireman, M., 1952, Exchangeable cation analysis of saline and alkali soils, *Soil Sci.*, 73: 251-261.

Gupta, R. K., Singh, C. P. and Abrol, I. P., 1985. Determination of CBC and exchangeable sodium in alkali soils. Soil Sci., 139: 326-332.

Richard, L. A., 1968, Diagnosis and improvement of saline water and alkali soils. USDA Agril, Hand Book No. 60. New Delhi, Oxford and IBH. pp. 7-33.

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Nutrient Composition of Different Animal Organic Manures

Different animal organic manures have been used in agriculture from time immemorial for maintaining and improving the fertility and productivity of soils. Nutrient content of these manures varies depending upon the types of feed used, age of the animal, and source of feed etc. Therefore, the study was taken to know the nutrient composition of some manures which are commonly used in preparing farm yard manure (FYM) by farmers.

Excreta of different animals, cattle dung slurry which is used for biogas production in UAS Campus, Dharwad and the digested slurry which comes out after the production of methane gas from the Gas Chamber are collected and dried under shade at room temperature. Dried samples were powdered in agate pestle and morter and passed through I mm sieve. Sieved material was digested in tri-acid digestion mixture (HNO₈:H₂SO₄: HClO₄ 10:1:4 ratio) and analysed for the macro, secondary and micronutrient contents by adopting standard procedures as outlined by Jackson (1967) and Piper (1950). The results are presented in Table 1.